## NOBLE GASES IN THE HAMLET METEORITE (LL4).

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**Introduction:** Heavy noble gases in primitive meteorites are contained in a very small portion of meteorites. This phase, Q for quintessence, is most likely carbonaceous [1, 2]. However, the exact nature of Q remains enigmatic. In continued effort to better understand Q [3-6], we carried out noble gas studies of the Hamlet meteorite (LL4), both for a bulk sample and an HF-HCl residue.

**Experimental:** The total of 3.47 g of the meteorite was processed at Washington University in St. Louis, USA. It was alternately treated with HF-HCl and HCl to dissolve silicates. Elemental sulfur was removed with CS<sub>2</sub>. The residue comprised 1.11% of the starting material. Noble gases were analyzed using the VG5400 at Osaka University, Japan. The temperatures for step-wise heating were 600, 800, 1000, 1200, 1400, and 1600°C for both samples.

**Results and Discussion:** The <sup>84</sup>Kr and <sup>132</sup>Xe concentrations in the bulk sample are  $1.24 \times 10^{-9}$  and  $1.40 \times 10^{-9}$  cm<sup>3</sup>STP/g, respectively. They are 2.2 and 2.8 times higher than those of the same meteorite by Alaerts et al. [7]. Helium-4, <sup>22</sup>Ne and <sup>36</sup>Ar concentrations are comparable in the two studies, indicating Q is more abundant in our bulk sample.

The <sup>132</sup>Xe concentration of the HF-HCl residue is  $9.24 \times 10^{-8}$  cm<sup>3</sup>STP/g, indicating that Q was enriched 66 times from the bulk sample. The mass balance calculation indicates that 73 percent of the Xe in the bulk sample remains in the HF-HCl residue. The Xe isotopic composition of the HF-HCl residue indicates that the Xe is essentially Xe-Q and that no diamond was in the residue.

The Ne in the bulk sample is dominated by cosmogenic Ne: the  ${}^{20}\text{Ne}/{}^{22}\text{Ne}$  and  ${}^{21}\text{Ne}/{}^{22}\text{Ne}$  ratios of the total are 0.812 ± 0.004 and  $0.866 \pm 0.003$ , respectively. The residue, although all silicates were removed, still contains a large amount of cosmogenic Ne  $(2.467 \pm 0.017 \text{ and } 0.6576 \pm 0.006, \text{ respectively})$ , indicating that it contains a significant amount of oxides. In order to determine the <sup>20</sup>Ne/<sup>22</sup>Ne ratio of Hamlet Ne-Q, we assumed 0.029 as the <sup>21</sup>Ne/<sup>22</sup>Ne ratio of Hamlet Ne-Q because <sup>21</sup>Ne/<sup>22</sup>Ne ratios of Ne-Q from many meteorites are known to be  $\sim 0.029$  [4, 8]. The  $^{20}$ Ne/ $^{22}$ Ne of Hamlet Ne-Q was determined to be 11.0 ± 0.5. <sup>20</sup>Ne/<sup>22</sup>Ne ratios of Ne-Q vary in different meteorites: they can be divided into two groups [8, 9]. The first group has <sup>20</sup>Ne/<sup>2</sup>Ne ratios of 10.11 ± 0.04, including Ne-Q from Lancé (CO3.5), and Cold Bokkeveld (CM2), and the second group has those of 10.57 ± 0.19, including Ne-Q from Allende (CV3), Chainpur (LL3.4), Grosnaja (CV3) and Murchison (CM2) [8]. Hamlet Ne-Q obviously falls into the second group.

**References:** [1] Reynolds J. H. et al. 1978. *Geochim. Cosmochim. Acta*, 42:1775-1797. [2] Ott U. et al. 1981. *Geochim. Cosmochim. Acta*, 45:1751-1788. [3] Matsuda J. et al. 2010. *Geochem. Cosmochim. Acta*, 74:5398-5409. [4] Matsuda J. et al. 2010. *Meteorit. Planet. Sci.*, 45:361-372. [5] Matsuda J. et al. 2010. *Meteorit. Planet. Sci.*, 45:1191-1205. [6] Amari S. et al. 2013. *Astrophys. J.*, 778:37 (39pp). [7] Alaerts L. et al. 1979. *Geochim. Cosmochim. Acta*, 43:1399-1415. [8] Busemann H. et al. 2000. *Meteorit. Planet. Sci.*, 35:949-973.[9] Huss G. R. et al. 1996. *Geochim. Cosmochim. Acta*, 60:3311-3340.